

$$m_{\text{shell}} = .400 \text{ kg}$$

$$r_{\text{shell}} = \frac{.200 \text{ m}}{2} = .100 \text{ m}$$

$$\rho_{\text{al}} = 806 \text{ kg/m}^3$$

Find Volume:

$$V = \frac{4}{3} \cdot \pi \cdot r^3$$

$$= \frac{4}{3} \cdot \pi \cdot (.100 \text{ m})^3$$

$$V_{\text{shell}} = 4.189 \times 10^{-3} \text{ m}^3$$

Find alcohol mass:

$$\rho_{\text{al}} = \frac{m_{\text{al}}}{V_{\text{al}}}$$

$$\text{so } 806 \text{ kg/m}^3 = \frac{m_{\text{al}}}{(4.189 \times 10^{-3} \text{ m}^3)}$$

$$m_{\text{al}} = 3.38 \text{ kg}$$

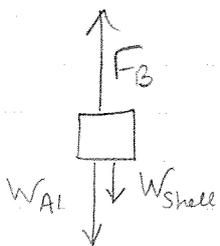
Find  $F_B$ :

$$F_B = \rho \cdot g \cdot V$$

$$= (1000 \text{ kg/m}^3)(9.8 \text{ m/s}^2)(4.189 \times 10^{-3} \text{ m}^3)$$

$$F_B = 41.05 \text{ N}$$

Find  $a$ :



$$F_B = 41.05 \text{ N}$$

$$F_{\text{net}} = m \cdot a$$

$$W_{\text{al}} = m \cdot g = (3.38 \text{ kg})(9.8 \text{ m/s}^2) = 33.09 \text{ N}$$

$$W_{\text{al}} = 33.09 \text{ N}$$

$$(41.05 - 33.09 - 3.92 \text{ N}) = (3.38 + .4 \text{ kg}) \cdot a$$

$$W_{\text{shell}} = m \cdot g = (.400 \text{ kg})(9.8 \text{ m/s}^2)$$

$$W_{\text{shell}} = 3.92 \text{ N}$$

$$a = 1.07 \text{ m/s}^2$$