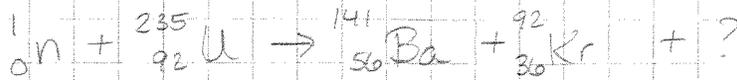


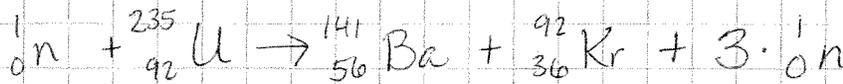
P # J

Ch 29- Worksheet



Atomic # (Z): $0 + 92 = 56 + 36$ so no extra protons are needed

Mass # (A): $1 + 235 = 141 + 92 + ?$ so 3 extra neutrons are needed



3 neutrons released