

$$(a) \quad m = 89.907737 \text{ u} \cdot \frac{1.66 \times 10^{-27} \text{ kg}}{1 \text{ u}}$$

$$m = 1.49 \times 10^{-25} \text{ kg}$$

$$(b) \quad {}_{38}^{90} \text{Sr} \quad \# \text{ of protons} = Z$$

$$Z = 38$$

$$(c) \quad {}_{38}^{90} \text{Sr} \quad \# \text{ of neutrons} = A - Z$$

$$= 90 - 38$$

$$\# \text{ neutrons} = 52$$